Palladium-Catalyzed Tandem Intramolecular Oxy/Amino-Palladation/Isocyanide Insertion: Synthesis of α-Benzofuranyl/Indolylacetamides

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Ⓢ *Supporting Information*

ABSTRACT: A novel palladium-catalyzed approach to 2-benzofuranyl/indolylacetamides from 1-(*o*-hydroxy/amino-phenyl)propargylic alcohols and isocyanides is described. The reaction proceeds through a cascade that includes oxy/aminopalladation, isocyanide insertion, and 1,4-hydroxyl migration. No oxidant or ligand is needed to promote the cascade, and the reactions are carried out under mild conditions to afford the products through high functional tolerance.
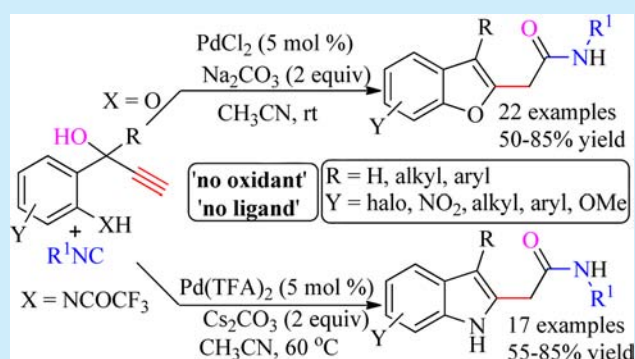
Isocyanide insertion has taken a new turn, for its reemergence, in recent chemistry with the assistance of palladium catalysis. Recent works by the groups of Wu,[1] Jiang,[2] Orru,[3] and others[3] for the synthesis of a variety of cyclic and acyclic motifs ensure a promising future for the profound usage of isocyanide as one carbon unit for a facile and simultaneous Pd-catalyzed stitching of two new and valuable C—C or C—heteroatom bonds. In continuation of our interest in the electrophilic cyclization of functionalized alkynes,[4] we herein report the synthesis of α-benzofuranyl/indolylacetamides via tandem intramolecular oxy/aminopalladation and isocyanide insertion. Benzofuran and indole derivatives have been considered among the highly privileged scaffolds for various medicinal chemistry aspects.[5] Synthesis of these moieties with diverse substitution patterns has always been one of the top priorities of chemists.

Earlier, Gabriele et al. reported a novel conversion of **1** to the α-benzofuranylacetamides **2** through CO insertion (Scheme 1).[6] The reaction required high pressure (30 atm) and temperature (100 °C), rendering it a specific benchtop reaction. Moreover, allylation was necessary for the initial palladium coordination, and the reaction was applicable for internal alkynes only (to produce branched substitution), whereas terminal alkynes adopted a different mode of cyclization.[7] Herein, we report a facile conversion of various nonallylated hydroxyphenyl propargyl alcohols **3** with a terminal alkyne to the corresponding α-benzofurnaylacetamides **4** at room temperature using PdCl₂ as the catalyst where no assistance of any ligand or additive was required (Scheme 1). Additionally, various aminophenyl propargyl alcohols **5** were

Scheme 1. Synthesis of α-Benzofuranyl/Indolylacetamides via Electrophilic Cyclization
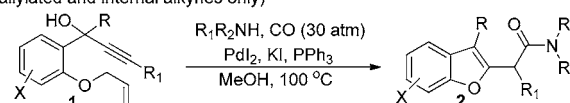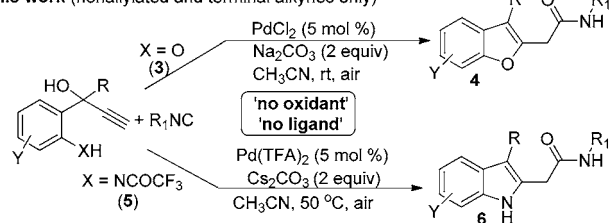
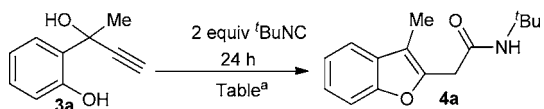
converted to the α-indolylacetamides **6** under slightly modified conditions.

Initially, we used the conditions, for the conversion of **3a** to **4a** (Table 1) via palladium-catalyzed tandem cyclization and isocyanide insertion, developed by Zhu et al.[3h] for the synthesis of indole-3-carboxamide via isocyanide insertion. Pleasingly, the expected product **4a** was obtained in 55% yield through a cascade of reactions (entry 1). Inspired by this result, some reaction parameters such as catalyst, solvent, and base effects were investigated for the purpose of improving the yield. Other

Table 1. Optimization of Reaction Conditions[a]

entry	catalyst	base	solvent	yield[b] (%)
1	PdCl₂	Na₂CO₃	DMSO	55
2	Pd(OAc)₂	Na₂CO₃	DMSO	30
3	Pd(TFA)₂	Na₂CO₃	DMSO	22
4	PdCl₂	Na₂CO₃	DCE	48
5	PdCl₂	Na₂CO₃	toluene	35
6	**PdCl₂**	**Na₂CO₃**	**CH₃CN**	**85**
7	PdCl₂	Na₂CO₃	1,4-dioxane	70
8	PdCl₂	K₂CO₃	CH₃CN	40
9	PdCl₂	CS₂CO₃	CH₃CN	20
10	PdCl₂	morpholine	CH₃CN	65
11	PdCl₂	TEA	CH₃CN	45
12	PdCl₂		CH₃CN	

[a]Reaction conditions: *t*BuNC (2.0 mmol), 3a (1.0 mmol), base (2.0 mmol) Pd(II)-cat. (0.05 mmol), solvent (4 mL), rt, open air. [b]Isolated yield.

catalysts like Pd(OAc)₂ and Pd(TFA)₂ under the same conditions were found to be less productive (entries 2 and 3). When various solvents were screened against PdCl₂ catalyst (entries 4–7), CH₃CN was found to produce the best yield (85%) of the product. Change of bases (entries 8–11) only led to decrease the yield. No product was obtained in the absence of the base (entry 12). Notably, all the reactions were carried out in open-air conditions. Interestingly, the reaction proceeded equally well in anhydrous/inert conditions, suggesting that there is no role of O₂ in the regeneration of the catalyst (vide infra, see mechanism).

With the optimized conditions in hand, we evaluated the generality of the method with respect to propargyl alcohols as well as isocyanides. A variety of hydroxyphenyl propargyl alcohols 3 were synthesized following the literature precedents.[7,9] As is evident from Scheme 2, a variety of substituents like alkyl, aryl, halogen, methoxy, and nitro groups were tolerated in the reaction to afford the products in moderate to high yields. Initially, various 3°-alcohols, the substrates prepared from *o*-hydroxyacetophenones and *o*-hydroxybenzophenones, were subjected to the reaction to obtain the corresponding products with a substitution at C3. Thus, 3,6-dimethylbenzofuranylacetamide 4b was obtained in high yield (82%), similar to product 4a. The presence of a bromo group as in the case of substrate 3c did not affect the outcome, producing 4c in 74% yield. Electron-poor substrate 3d was found to react equally well (to yield 4d in 82%), whereas its electron-rich counterpart 3e produced the corresponding product 4e in a moderate yield of 52%. Similarly, 3-phenyl-substituted benzofuran adducts 4f–h were obtained from 3°-alcohols 3f–h in comparative yields (75–81%). Then, various electron rich, neutral and poor 2°-alcohols, prepared from salicylaldehydes, were subjected to the cascade reaction. Delightedly, they all produced the corresponding products (4i–m) although in relatively slightly lower yields (50–72% compared to 52–82%). Next, we aimed to evaluate the reactivity of a variety of commercially available isocyanides. Thus, isopropyl, tetramethylbutyl, and cyclohexyl isocyanides were subjected to the transformation. All of them smoothly produced the corresponding products in 63–75% yields, although the relative productivity was less compared

**Scheme 2. Synthesis of α-Benzofuranylacetamides 4 via Oxypalladation and Isocyanide Insertion**[a]
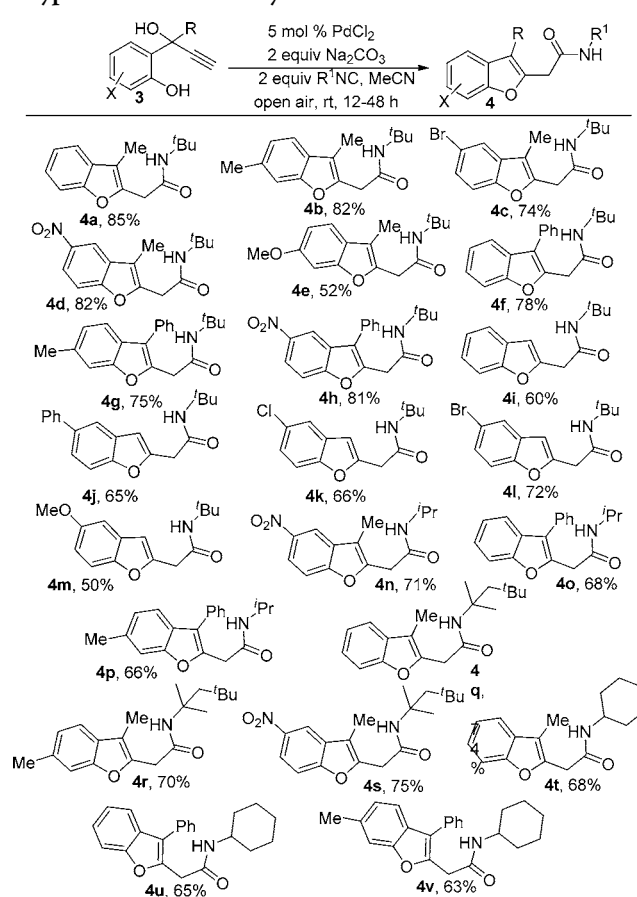
[a]Reaction conditions: R¹NC (2.0 mmol), 3 (1.0 mmol), base (2.0 mmol), PdCl₂ (0.05 mmol), MeCN (0.2 M), open air, rt.

with *tert*-butyl isocyanide. Unfortunately, phenyl isocyanide was found to be unreactive under the given conditions. This diminished reactivity and inertness of less hindered and aryl isocyanides, respectively, is in accordance with the recent observations by Ji et al.[3f] Setting a limitation, the reaction did not work for the internal alkyne, which produced a non-isocyanide-inserted product.[8]

Our curiosity next turned toward the extension of this reaction to 2-aminophenyl propargyl alcohols 5,[10] which might give valuable indole counterparts 6. Under the standardized conditions described above, the corresponding product (6) was obtained only in traces. After some optimization studies,[11] we found that 5 mol % of Pd(TFA)₂ in the presence of 2 equiv of Cs₂CO₃ in acetonitrile with a slightly elevated temperature (60 °C) allowed the transformation with reasonable to high yields (Scheme 3). Thus, 3-phenyl-2-indolylacetamide (6a) and its 5-methyl adduct (6b) were obtained in 72% and 70% yields, respectively. Then a variety of the substrates were synthesized from various commercially or readily available 2-amino-acetophenones and 2-aminobenzophenones[10] to evaluate the generality of the method.

Halogen groups on the core reaction part (5c) as well as on the pendant phenyl ring (5d–j) were well tolerated in the reaction to produce the corresponding products 6c–j in yields ranging from 65% to 85%.

Methoxy-substituted substrate 5k also smoothly transformed to the indolylacetamide 6k in 80% yield. The 3-methyl-2-

Scheme 3. Synthesis of α-Indolylacetamides 6 via Aminopalladation and Isocyanide Insertion[a]
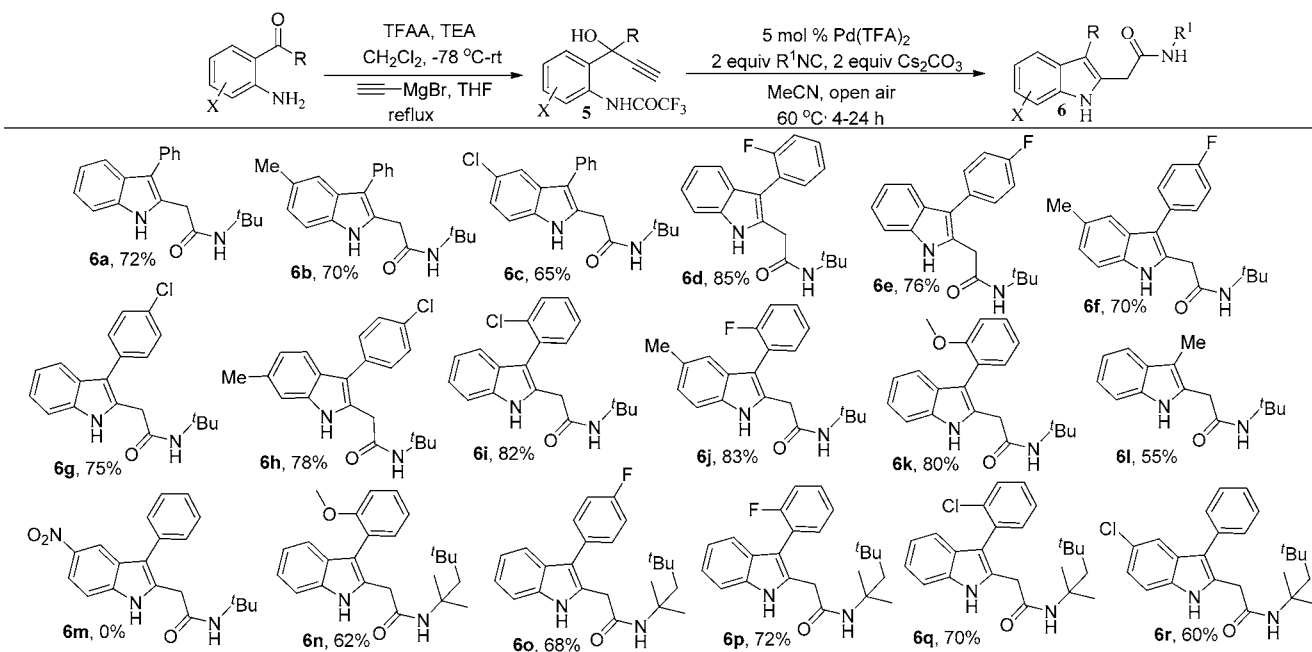
[a]Reaction conditions: R[1]NC(2.0 mmol), **5** (1.0 mmol), base (2.0 mmol), Pd(TFA)$_2$ (0.05 mmol), MeCN (0.3 M), open air, 60 °C.

indolylacetamide **6l** was obtained in moderate yield of 55%, which can be attributed to the instability of the substrate **5l** that could not be purified but was used as the crude product after the Grignard reaction. Surprisingly, electron-poor substrate **5m** was found to be unreactive in the transformation, which is in sharp contrast to its benzofuran counterpart (**4d**, **4h**, **4n**, and **4s**) synthesis. Next, tetramethylbutyl isocyanide was used as the reaction partner with various halo-substituted and methoxy-substituted substrates to obtain the corresponding products **6n**−**r** in 62−72% yields. Unfortunately, less hindered isopropyl isocyanide and phenyl isocyanide were found to be unreactive in the reaction. In addition, the internal alkyne found to be inert in the reaction even after the prolonged reaction time and at the elevated temperatures.[8]

The structures of all the products were characterized by ^{1}H NMR, ^{13}C NMR and HRMS. For unambiguous structure confirmation, we have taken single-crystal X-ray structures of **4b** and **6j**[12] (one from each group of products) which are depicted in Figure 1.

A plausible mechanism for the cascade of reactions in the above transformations is described in Figure 2. We speculate that the palladium catalyst shuttles between two oxidation states (2 and 0) in each cycle, which helps effective regeneration of the catalyst without the use of any oxidant. Accordingly, activation of alkyne moiety by PdX$_2$ may lead to oxy/amino palladation in *anti* and *5-exo-dig* fashion to give intermediate **A**. Isocyanide insertion (**B**) followed by coordination palladium with spatially closer hydroxyl group would form 6-membered oxapalladacycle **C**. Reductive elimination of Pd(0) followed by its immediate insertion in to allylic ether of **D**, a well-established phenomenon, in the newly formed strained 5-membered oxacycle may give 6-membered oxapalladacycle **E** which on protodepalladation by two molecules of HX (generated in the cycle) would lead to the formation of intermediate **F**. The concomitant 1,3-H shift and tautomerization of the iminol group would furnish **4** and protected **6** (**G**). The sensitive trifluoroacetyl group in **6** would
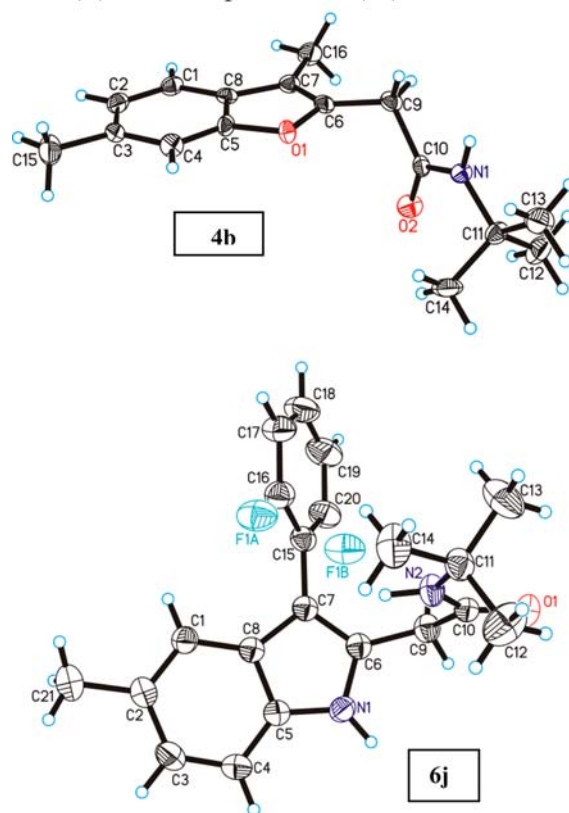

Figure 1. X-ray crystal structures of **4b** (CCDC No. 978848) and **6j** (CCDC No. 996275).[12]

subsequently undergo hydrolysis in the presence of Pd(II) with the help of moisture (air) to liberate **6**.

In conclusion, a novel and efficient palladium-catalyzed intramolecular oxy/aminopalladative isocyanide insertion approach for the synthesis of 2-benzofuranyl/indolylacetamides is
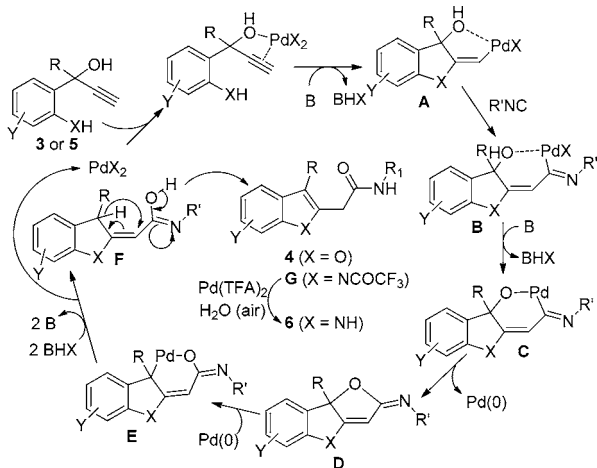
Figure 2. Proposed catalytic cycle.

demonstrated. The approach offers advantages in terms of experimental simplicity, mild reaction conditions, and broad functional group tolerance.

ASSOCIATED CONTENT

Supporting Information

Experimental details and copies of ^1H and ^{13}C spectra for compounds **4a–v**, **5**, and **6a–r**. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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(11) For optimization studies, see the Supporting Information.

(12) The ORTEP diagram of **6j** in Figure 1 shows an average structure of two atropisomers, and hence, the F atom is shown 50% each on both ortho positions of the pendant phenyl ring.